MICROBIAL TRANSFORMATIONS OF 6α- AND 6β-EUDESMANOLIDES BY *RHIZOPUS NIGRICANS* CULTURES

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ABSTRACT.—Microbial transformations of 6α -eudesmanolides and 6β -eudesmanolides were accomplished using the hydroxylating fungus *Rhizopus nigricans*. The eudesmane skeleton was functionalized principally at positions C-1, C-4, and C-8. Hydroxylation yields depended on the configuration of the hydroxyl group at C-3, the methyl group at C-4, and the lactone ring in the substrates. Several metabolites with hydroxylation at C-9 and C-11 were also obtained with 6α -eudesmanolide substrates. An inversion of configuration at C-3 may occur when the substrates have an axial hydroxyl group at this carbon. Hydroxylation at C-4 was observed only when the substrate had a 4β -methyl group.

Sesquiterpene lactones with a eudesmane- 6α ,12-olide structure are abundant in nature and the chemistry (1–7) and biotransformation (8–14) of some have been studied extensively. However, 6β -eudesmanolides are scarce and have rarely been studied. As a continuation of our studies on the systematic biotransformation of terpenes (15–20), we present the results of microbial transformations of some 6α -and 6β -eudesmanolides using the fungus *Rhizopus nigricans*. Bioconversion permits access to remote positions on the molecules and is an alternative to chemical routes. The data obtained in this work, together with previous findings (17,18), help to establish the relationship between the structures of eudesmanolide substrates and the hydroxylating activity of this fungus. We describe herein a series of biotransformations on commercial α -santonin. These conversions have yielded eudesmanolides hydroxylated at C-1 that can be chemically transformed into other sesquiterpene skeletons. We have also obtained 6-eudesmanolides with hydroxylation at C-8 which can be converted to 8-eudesmanolides with either an 8α - or 8β -configuration.

RESULTS AND DISCUSSION

The hydrogenation of α -santonin [1] was described many years ago (21,22). We reduced commercial α -santonin [1] (H₂/PtO₂), yielding products 2 (21) (75%) and 3 (21) (20%) (Scheme 1). ¹H- and ¹³C-nmr spectra of 2 indicated that the hydroxyl group in this compound was located at C-3 and the methyl group was at C-4, both with a β -configuration. Therefore, 2 was formed by hydrogenation of the A ring on the α -face of the molecule. Product 3 exhibited a C-3 geminal proton signal (ddd, $J_1=J_2=J_3=2.8$ Hz), which indicated that it was equatorial. The methyl group at C-4 in this compound [3] had a β -configuration (J=7.6 Hz).

Partial hydrogenation of **1**, followed by treatment with acid and NaBH₄ reduction gave rise to compounds **4** (23) (74%) and **5** (23) (20%) (Scheme 1). These products are epimers at C-3 with the methyl group at C-4 α . The compound with a 3 β -hydroxyl configuration was again the main product.

Epimerization of α -santonin [1] at C-6 to 6 β -santonin [6] was carried out as previously described (17) (Scheme 1). Catalytic hydrogenation of 6 with H₂/PtO₂ yielded the previously described products 7 (80%) (18) and 8 (15%) (17).



SCHEME 1. Starting materials 2, 4, 5, 7, and 9 obtained from $(-)-\alpha$ -santonin [1].

Hydrogenation of the A ring in 6β -santonin [6] with H₂/Pt/C gave a C-3 carbonyl compound that was epimerized at C-4 in acidic medium (17). Finally, NaBH₄ reduction of this ketone yielded compounds 8(75%) and 9(20%)(17). Nmr experiments indicated that 9 had a 3 α -hydroxyl group and a 4 α -methyl group.

 3β -Hydroxy- 4α , 5α , 11β -H-eudesman- 6α , 12-olide [2] was incubated with *Rhizopus nigricans* for 12 days to give the unaltered substrate 2 (8%), metabolite 10 (32%), metabolite 11 (18%), and a mixture of more polar metabolites (Scheme 2). Acetylation of this mixture of metabolites gave the products 12 (4%), 13 (7%), 14 (10%), and 15 (2%). The cims of 10 showed a molecular peak at m/z 269 ([M+1]⁺), indicating that 10 had a new hydroxyl group compatible with a signal at δ 3.37 (1H, dd, J_1 =11.5 Hz, J_2 =4.2 Hz) in its ¹H-nmr spectrum. On the basis of this signal and the ¹³C-nmr spectrum of 10, we propose that the hydroxylation occurred at C-1. The new hydroxyl-group configuration was confirmed by nOe experiments. Irradiation at H-3 produced a nOe effect on H-1, thus metabolite 10 was a 1 β -hydroxyl derivative.

The second metabolite [11] of this incubation also had a cims molecular peak at m/z 269. The C-15 methyl group signal of 11 was converted to a singlet at δ 1.25 (3H, s), and the H-3 signal was at δ 3.52 (1H, dd, J_1 =11.8 Hz, J_2 =4.6 Hz). Therefore, the new hydroxyl was at C-4, a conclusion corroborated by its ¹³C-nmr spectrum, where a new oxygenated carbon (δ 75.0) was detected. Several nOe difference experiments were performed to determine the configuration of 11 at C-4. Irradiation at δ 4.05 (H-6 β) produced a nOe on the C-15 methyl group, and vice versa, indicating an α -configuration.

Product **12** was identified as the acetate of **2**. Acetate **13** corresponded to the C-3 monoacetate of metabolite **11**. Compound **14** had a cims molecular peak of m/z 353, which indicated that it was a diacetate of **2**. Geminal protons of the two acetoxyl groups appeared in its ¹H-nmr spectrum at δ 4.81 (1H, ddd, $J_1 = 10.7$ Hz, $J_2 = J_3 = 5.2$ Hz, H- 3α) and δ 5.08 (1H, ddd, $J_1 = 15.1$ Hz, $J_2 = 10.9$ Hz, $J_3 = 4.2$ Hz), respectively. These observations, together with the ¹³C-nmr spectrum, indicated that the new acetoxyl group was at C-8 with an 8S configuration. This 8 α -arrangement of the acetoxyl group

was confirmed by nOe experiments between the H-8, H-11, and Me-14 protons. The last product [15], obtained in low yield, had a single acetoxyl group at C-3 and a new hydroxyl group. Its ¹H-nmr spectrum showed two singlet methyl groups at δ 1.05 and δ 1.24; the chemical shift and the multiplicity of these methyl group signals indicated that the new hydroxyl group was situated at C-11. This conclusion was confirmed by the presence of a new quaternary oxygenated carbon in its ¹³C-nmr spectrum.

Incubation of substrate 4 with *Rhizopus nigricans* for 10 days gave metabolites **16** (40%), **17** (7.5%), **18** (16%), and **19** (13%) (Scheme 2). The main metabolite [**16**] isolated from this biotransformation showed ¹H-nmr signals at δ 3.37 (1H, dd, J_1 =11.9 Hz, J_2 =4.4 Hz). An α -effect for C-1 and β -effects for C-2 and C-10 were observed in its ¹³C-nmr spectrum. Therefore, metabolite **16** had a new equatorial hydroxyl group at C-1.

Metabolite **17** gave a cims molecular peak of m/z 269; its ¹H-nmr spectrum showed a signal at δ 3.94 (1H, ddd, J_1 =14.9 Hz, J_2 =10.6 Hz, J_3 =4.4 Hz) due to an axial proton geminal to a new hydroxyl group at C-8. This hydroxylation was confirmed by the ¹³Cnmr data: an α -effect on C-8 ($\Delta\delta$ +45.6), β -effects on C-7 ($\Delta\delta$ +6.1) and C-9 ($\Delta\delta$ +10.5), and γ -effects on C-6 ($\Delta\delta$ -2.8) and C-10 ($\Delta\delta$ -0.5) were observed.

The third metabolite [18] obtained from this incubation had spectroscopic data very similar to those of metabolite 16. The ¹H-nmr spectrum of 18 indicated that it had an axial proton geminal to a new hydroxyl group (δ 3.37, 1H, dd, J_1 =10.2 Hz, J_2 =4.2 Hz). The multiplicity and coupling constants of this signal indicated that the new hydroxyl group was at C-9 with a 9 β -configuration. ¹³C-Nmr data of 18 corroborated this conclusion; thus, an α -effect on C-9 ($\Delta\delta$ +37.3), β -effects on C-8 ($\Delta\delta$ +6.8) and C-10 ($\Delta\delta$ +4.7), and γ -effects on C-1 ($\Delta\delta$ -4.0), C-5 ($\Delta\delta$ -2.2), C-7 ($\Delta\delta$ -2.7) and C-14 ($\Delta\delta$ -3.8) were noted.

The last metabolite obtained from this biotransformation [19] had a cims molecular peak of m/z 269, which indicated that substrate 4 had again been hydroxylated by the fungus. The ¹H-nmr spectrum of 19 indicated that the new hydroxyl group was introduced at a tertiary carbon. The hydroxyl group was therefore expected at C-4 or C-11. Comparison of the ¹³C-nmr chemical shifts of C-11, C-6, C-7, and C-8 to those of substrate 4 showed that the new hydroxyl group was at C-11. Finally, the OH-11 configuration was determined by several nOe experiments. Irradiation on the C-13 methyl group produced a nOe effect with H-6 β and vice versa; therefore the hydroxyl group at C-11 had an 11 β -configuration.

When substrate **5** was incubated with *Rhizopus nigricans* for 10 days, only two metabolites (**16**, 30%, and **20**, 17%) were isolated. The first metabolite of this biotransformation had physical and spectroscopic properties identical to those of metabolite **16**, isolated previously from the incubation of substrate **4**. Therefore, the microorganism had functionalized at C-1 and epimerized at C-3. Metabolite **20** was a dihydroxylated eudesmanolide (molecular mass 268 daltons). Comparison of the ¹³C-nmr spectrum of compound **5** and metabolite **20** showed that the new functionalization was at C-9. On the basis of these observations, the structure 3α ,9 β -dihydroxy-4 β ,5 α ,11 β -H-eudesman-6 α ,12-olide was proposed for metabolite **20**.

Biotransformation of substrate 7 with *Rhizopus nigricans* for 12 days gave metabolites **21** (16%), **22** (20%), **23** (12%), and a mixture of more polar metabolites (Scheme 2). This mixture was acetylated, and products **24** (4%), **25** (10%), and **26** (5%) were isolated. Metabolite **21** showed spectroscopic behavior that indicated the presence of a new hydroxyl group at C-1, with a 1 β configuration (δ 3.21, 1H, dd, J_1 =11.5 Hz, J_2 =4.3 Hz). The physical and spectroscopic data of metabolite **22** (4 α -hydroxyl derivative) were consistent with those reported previously for this compound (18). The



SCHEME 2. Biotransformation results of substrates 2, 4, 5, 7, and 9.

¹H-nmr spectrum of metabolite **23** showed a new signal of a geminal hydroxyl group proton (δ 3.53, 1H, ddd, J_1 =12.0 Hz, J_2 =9.9 Hz, J_3 =4.2 Hz). The multiplicity and the coupling constants of this signal indicated that the new hydroxylation was situated at C-2 or at C-8. A comparison of the ¹³C-nmr chemical shifts of metabolite **23** to those of substrate **7** revealed that this compound had an 8 α -hydroxyl group (β -effects on C-7 and C-9, $\Delta\delta$ +9.2 and +10.3, respectively). As shown in Scheme 2, metabolites **24**, **25**, and **26** were the acetylated derivatives of **21**, **22**, and **23**, respectively, and were identified by their physical and spectroscopic data.

Finally, we carried out the bioconversion of the 6β -eudesmanolide 9 with *Rhizopus* nigricans for 7 days. Metabolites 8 (17%), 27 (31%), and a mixture of additional metabolites were isolated. Acetylation of this mixture gave the acetylated derivatives 28 (4%), 29 (4%), and 30 (17%). The first metabolite isolated from this biotransformation had physical and spectroscopic properties identical to those of 8. This metabolite was formed by epimerization of substrate 9 at C-3. The structure of metabolite 27 was deduced by comparison of its spectroscopic data with those described for product 23, which confirmed a new hydroxylation at C-8 with an 8 α -configuration.

Product **28** had a cims that indicated a new acetoxyl group (molecular peak of m/z 311). Taking into account the previously described results, and the ¹H- and ¹³C-nmr spectra of **28**, we could establish that this metabolite had a 1 β -acetoxyl group that was due to a 1 β -hydroxylation of substrate **9** by the fungus and subsequent partial acetylation by chemical means. The diacetylated compound **29** was identified as the acetoxyl derivative of **28**. Product **30** had spectroscopic data indicating that it was a

diacetoxyl eudesmanolide that could be formed by epimerization at C-3 of the hydroxyl group in this position of substrate 9, and subsequent 1β -hydroxylation.

Several conclusions can be drawn from the above results of the 6α - and 6β eudesmanolide biotransformations with *Rhizopus nigricans* and from our previously published results (17,18). Thus, *Rhizopus nigricans* functionalizes the C-1, C-4, and C-8 eudesmane skeleton positions, although the yields depend on the C-3, C-4, and C-6 substituent arrangements. Epimerization at C-3 was observed when the hydroxyl group at this position had an α -configuration. If the substrate was a 6α -eudesmanolide, minor hydroxylations at C-9 and C-11 were observed. The microorganism *Rhizopus nigricans* hydroxylated at C-4 only when the methyl group at this position was axial (β configuration). Some of these biotransformations give acceptable yields of C-1 or C-8 hydroxylated products, which are adequate starting materials to attempt the biogenetic formation of guaianolides and 8,12-eudesmanolides.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—Mps were determined using a Kofler (Reichter) apparatus and are uncorrected. Optical rotations were measured on a Perkin-Elmer 141 polarimeter at 20°. Ir spectra were recorded on a Perkin-Elmer 983G spectrometer or on a Nicolet 20SX Ft-ir spectrometer. Ms were determined with ci (CH₄) using a Hewlett-Packard 5988A mass spectrometer. Elemental analyses were made in a Perkin-Elmer 240C analyzer. Nmr spectra (400.13 MHz¹H, 300.13 MHz¹H, and 100.62 MHz¹³C) were recorded in CDCl₃ (which also provided the lock signal) on Bruker ARX-400 and Bruker AM-300 spectrometers. The assignments of ¹³C-nmr chemical shifts were done with the aid of DEPT experiments using a flip angle of 135°. Si gel SDS 60A CC (40–60 μ m) was used for flash chromatography. CH₂Cl₂ or CHCl₃ containing increasing amounts of Me₂CO were used as the eluent. Analytical tlc plates (Si gel, Merck 60 G) were visualized by spraying with H₂SO₄/AcOH, followed by heating to 120°. The starting material was (–)- α -santonin (99%, Aldrich Chemical Company).

ORGANISM, MEDIA, AND CULTURE CONDITIONS.—*Rhizopus nigricans* CECT 2072 was obtained from the Coleccion Española de Cultivos Tipo, Departamento de Microbiologia, Facultad de Ciencias, Universidad de Valencia, Spain, and was kept in YEPGA medium, containing yeast extract (1%), peptone (1%), glucose (2%), and agar (2%) in H₂O at pH 5. In all the transformation experiments a medium of peptone (0.1%), yeast extract (0.1%), beef extract (0.1%), and glucose (0.5%) in H₂O at pH 5.7 was used. Erlenmeyer flasks (250 ml) containing 80 ml of medium were inoculated with a dense suspension of *Rhizopus nigricans*. The cultures were incubated by shaking (150 rpm) at 28° from 6 days, after which substrates **2**, **4**, **5**, **7**, and **9** in EtOH (1%) were added.

CATALYTIC HYDROGENATION OF (-)- α -SANTONIN [1]. (-)- α -Santonin (2 g) was dissolved in EtOH (30 ml) and hydrogenated with H₂ (4 atmospheres) on PtO₂ (300 mg) for 5 h at room temperature. Si gel cc yielded 3 β -hydroxy-4 α ,5 α ,11 β -H-eudesman-6 α ,12-olide [2] (21) (1.53 g, 75%) and 3 α -hydroxy-4 α ,5 α ,11 β -H-eudesman-6 α ,12-olide [3] (21) (412 mg, 20%).

 3β -Hydroxy-4 α , 5α , 11β -H-eudesman- 6α , 12-olide [2].—Mp 125°; $[\alpha]^{25}$ D -62° (c=1, CHCl₃); ir ν max (CHCl₃) 3460, 1767, 1237 cm⁻¹; ¹H nmr (CDCl₃) δ 3.95 (1H, dd, $J_{6,5}=11.6$ Hz, $J_{6,7}=9.9$ Hz, H-6 β), 3.72 (1H, ddd, $J_{3,2\beta}=11.0$ Hz, $J_{3,4\alpha}=J_{3,2\alpha}=5.3$ Hz, H-3 α), 2.39 (1H, ddq, $J_{4\alpha,15}=7.5$ Hz, $J_{4\alpha,3}=J_{4\alpha,5}=5.3$ Hz, H-4 α), 2.33 (1H, dq, $J_{11,7}=12.2$ Hz, $J_{11,13}=6.9$ Hz, H-11 β), 1.19 (3H, d, $J_{13,11}=6.9$ Hz, Me-13), 1.01 (3H, s, Me-14), 0.96 (3H, d, $J_{15,4}=7.5$ Hz, Me-15); ¹³C nmr (CDCl₃) δ 40.2 (C-1), 25.9 (C-2), 73.0 (C-3), 34.4 (C-4), 49.7 (C-5), 80.1 (C-6), 53.6 (C-7), 23.5 (C-8), 43.3 (C-9), 35.9 (C-10), 41.9 (C-11), 178.7 (C-12), 12.6 (C-13), 21.1 (C-14), 8.8 (C-15); cims (CH₄) m/z 253 [M+1]⁺ (50), 235 [M+1-18]⁺ (100); anal., found C 71.3, H 9.6; C₁₅H₂₄O₃ requires C 71.38, H 9.59%.

 3α -Hydroxy-4 α , 5α , 11β -H-eudesman- 6α , 12-olide [**3**].—Mp 116–118°, $[\alpha]^{25}$ D – 54° (c=1, CHCl₃); ir $\nu \max(\text{CHCl}_3)$ 3460, 1767 cm⁻¹; ¹H nmr (CDCl₃) δ 3.93 (1H, dd, $J_{6,5}=11.6$ Hz, $J_{6,7}=10.0$ Hz, H-6 β), 3.85 (1H, ddd, $J_{3,4\alpha}=J_{3,2\alpha}=J_{3,2\beta}=2.8$ Hz, H-3 β), 2.32 (1H, dq, $J_{11,7}=12.4$ Hz, $J_{11,13}=6.9$ Hz, H-11 β), 2.10 (1H, m, H-4 α), 2.03 (1H, dd, $J_{5,6}=11.6$ Hz, $J_{5,4}=5.0$ Hz, H-5 α), 1.20 (3H, d, $J_{11,13}=6.9$ Hz, Me-13), 1.02 (3H, s, Me-14), 0.99 (3H, d, $J_{15,4}=7.6$ Hz, Me-15); cims (CH₄) m/z 253 [M+1]⁺ (49), 235 [M+1-18]⁺ (100); anal., found C 71.4, H 9.6; C₁₅H₂₄O₃ requires C 71.38, H 9.59%.

COMPOUNDS 4 AND 5 FROM $(-)-\alpha$ -SANTONIN [1].—A solution of $(-)-\alpha$ -santonin (2 g) in CH₂Cl₂(60 ml) was partially hydrogenated with H₂(4–5 atmospheres) for 5 h on Pt-charcoal. Then, the reaction mixture was filtered and concentrated. CHCl₃ (50 ml) and several drops of aqueous 2 N HCl were added and the

mixture allowed to stand for 7 days at room temperature. The reaction mixture was concentrated, dissolved in EtOH (200 ml), NaBH₄ (735 mg) added, and the solution stirred for 2 h at 0°. Finally the reaction was treated with dilute HCl, H₂O, and extracted thoroughly with CH₂Cl₂. The CHCl₃ solubles were chromatographed on Si gel and products **4** (23) (1.52 g, 74%) and **5** (23) (406 mg, 20%) were isolated.

CONVERSION OF (-)- α -SANTONIN [1] TO (-)-6-EPI- α -SANTONIN [6] BY THE ISHIKAWA PROCEDURE. (-)- α -Santonin (5 g) was dissolved in anhydrous N,N'-dimethylformamide (50 ml) containing anhydrous HCl (5%). The solution was heated for 3 h at 85°. The mixture was stirred at room temperature for 12 h, then diluted with H₂O and extracted with CH₂Cl₂. The organic layer was washed with saturated aqueous NaCl, saturated aqueous NaHCO₃, and water. The solvent was removed under reduced pressure. Chromatography over Si gel yielded (-)-6-epi- α -santonin (6, 4.2 g, 84%) (17).

CATALYTIC HYDROGENATION OF (-)-6-*EPI*- α -SANTONIN [6].—A solution of (-)-6-*epi*- α -santonin (1.5 g) in CH₂Cl₂ (50 ml) was hydrogenated for 6 h with H₂ (4–5 atmospheres) on PtO₂ (0.25 g). The reaction mixture was filtered and concentrated *in vacuo*. Chromatography on a Si gel column yielded 7 (18) (1.23 g, 80%) and **8** (17) (230 mg, 15%).

COMPOUNDS **8** AND **9** FROM (-)-6-*EPI*- α -SANTONIN [**6**].—A solution of (-)-6-epi- α -santonin (2.5 g) in CH₂Cl₂(80 ml) was partially hydrogenated with H₂(4–5 atmospheres) for 6 h on Pt-charcoal. The reaction mixture was dissolved in CHCl₃ (60 ml) and several drops of a 2 N HCl solution were added. The reaction mixture was shaken vigorously at room temperature for 3 days. The resultant mixture was treated with EtOH (300 ml), with NaBH₄ (900 mg) added, keeping the reaction for 2 h at 0°. The reaction was then treated with dilute HCl, diluted with H₂O, and extracted with CH₂Cl₂. The solvent was removed under reduced pressure and chromatographed on a Si gel column yielding **8** (17) (1.92 g, 75%) and **9** (17) (511 mg, 20%).

BIOTRANSFORMATION OF 3β -HYDROXY- 4α , 5α , 11β -H-EUDESMAN- 6α ,12-OLIDE [2].—Substrate 2 (725 mg) was dissolved in EtOH (15 ml), distributed among 15 Erlenmeyer-flask cultures and incubated for 12 days after which the cultures were filtered and pooled; the cells were washed thoroughly with H₂O and the liquid was saturated with NaCl and extracted twice with CH₂Cl₂. Extracts were pooled, dried (Na₂SO₄), and evaporated at 40° in vacuo to give a mixture of compounds (695 mg). This mixture was chromatographed on a Si gel column to obtain 60 mg (8%) of starting material, 250 mg (32%) of metabolite **10**, 140 mg (18%) of metabolite **11**, and 245 mg of a polar mixture, which was dissolved in Ac₂O/pyridine (1:2) and maintained for 12 h at room temperature. The reaction mixture was diluted with H₂O, extracted with CH₂Cl₂, washed with saturated aqueous KHSO₄ and dried with anhydrous Na₂SO₄. Chromatography on Si gel yielded the acetoxy derivatives **12** (35 mg, 4%), **13** (60 mg, 7%), **14** (100 mg, 10%), and **15** (20 mg, 2%). All yields are given with respect to substrate **2**.

 $\begin{array}{l} 1\beta, 3\beta - Dihydroxy-4\alpha, 5\alpha, 11\beta - H-eudesman-6\alpha, 12-olide \ [10], \\ --Mp \ 195^\circ; \ [\alpha]^{25}D + 12^\circ \ (c=1, CHCl_3); \\ \text{ir } \nu \ \text{max} \ (CHCl_3) \ 3424, 1768 \ \text{cm}^{-1}; \ ^1H \ \text{nmr} \ (CDCl_3) \ \delta \ 4.06 \ (1H, \ dd, J_{6,5}=11.4 \ Hz, J_{6,7}=10.0 \ Hz, H-6\beta), \\ 3.82 \ (1H, \ ddd, J_{3,2\beta}=10.2 \ Hz, J_{3,4\alpha}=5.7 \ Hz, J_{3,2\alpha}=4.6 \ Hz, H-3\alpha), 3.37 \ (1H, \ dd, J_{1\alpha,2\beta}=11.5 \ Hz, J_{1\alpha,2\alpha}=4.2 \ Hz, H-1\alpha), 2.38 \ (1H, \ ddq, J_{4\alpha,5}=13.7 \ Hz, J_{4\alpha,15}=7.5 \ Hz, J_{4\alpha,3\alpha}=6.8 \ Hz, H-4\alpha), 2.32 \ (1H, \ dq, J_{11,7}=13.9 \ Hz, J_{11,13}=7.1 \ Hz, H-11\beta), 1.22 \ (3H, \ d, J_{13,11}=7.1 \ Hz, Me-13), 1.02 \ (3H, \ s, Me-14), 0.98 \ (3H, \ d, J_{15,4}=7.5 \ Hz, Me-15); \ ^{13}C \ nmr \ (CDCl_3) \ \delta \ 77.6 \ (C-1), 35.2 \ (C-2), 70.6 \ (C-3), 34.3 \ (C-4), 47.6 \ (C-5), 79.5 \ (C-6), 53.7 \ (C-7), 23.3 \ (C-8), 39.7 \ (C-9), 41.3 \ (C-10), 42.0 \ (C-11), 179.5 \ (C-12), 12.6 \ (C-13), 15.2 \ (C-14), 8.7 \ (C-15); \\ \text{cims} \ (CH_4) \ m/z \ 269 \ \ [M+1]^+ \ (28), 251 \ \ \ [M+1-18]^+ \ (100), 233 \ \ \ \ [M+1-36]^+ \ (70); \ anal., \ found \ C \ 67.0, \ H \ 9.02\%. \end{array}$

 $3\beta,4\alpha$ -Dibydroxy- $5\alpha,11\beta$ -H-eudesman- $6\alpha,12$ -olide [**11**].—Syrup; [α]²⁵D +13° (c=1, CHCl₃); ir ν max (CHCl₃) 3464, 1781 cm⁻¹; ¹H nmr (CDCl₃) δ 4.05 (1H, dd, $J_{6,5}$ =11.3 Hz, $J_{6,7}$ =10.4 Hz, H-6 β), 3.52 (1H, dd, $J_{3,2\beta}$ =11.8 Hz, $J_{3,2\alpha}$ =4.6 Hz, H-3 α), 2.26 (1H, dd, $J_{1,1,7}$ =12.5 Hz, $J_{11,13}$ =6.9 Hz, H-11 β), 1.18 (3H, d, $J_{13,11}$ =6.9 Hz, Me-13), 1.25 (3H, s, Me-15), 0.96 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 39.1 (C-1), 26.0 (C-2), 77.4 (C-3), 75.0 (C-4), 55.6 (C-5), 80.6 (C-6), 53.2 (C-7), 23.3 (C-8), 42.9 (C-9), 37.6 (C-10), 40.4 (C-11), 178.3 (C-12), 12.3 (C-13), 19.9 (C-14), 17.8 (C-15); cims (CH₃) m/z 269 [M+1]⁺ (15), 251 [M+1-18]⁺ (100), 233 [M+1-36]⁺ (35); anal., found C 67.0, H 8.9; C₁₅H₂₄O₄ requires C 67.12, H 9.02%.

3β-Acetoxy-4α, 5α, 11β-H-eudesman-6α, 12-olide [12].—Mp 205°; [α]²⁵D +23° (c=1, CHCl₃); ir ν max (CHCl₃) 1766, 1734 cm⁻¹; ¹H nmr (CDCl₃) δ 4.79 (1H, ddd, $J_{3,28}$ =11.7 Hz, $J_{3,4α}$ =10.5 Hz, $J_{3,2α}$ =5.2 Hz, H-3α), 3.92 (1H, dd, $J_{6,5}$ =11.4 Hz, $J_{6,7}$ =9.9 Hz, H-6β), 2.50 (1H, ddd, $J_{4\alpha,3}$ =12.5 Hz, $J_{4\alpha,15}$ =7.4 Hz, $J_{4\alpha,5}$ =5.1 Hz, H-4α), 2.31 (1H, dq, $J_{11,7}$ =12.2 Hz, $J_{11,13}$ =6.9 Hz, H-11β), 2.01 (3H, s, Me-AcO), 1.19 (3H, d, $J_{13,11}$ =6.9 Hz, Me-13), 1.03 (3H, s, Me-14), 0.96 (3H, d, $J_{15,4}$ =7.4 Hz, Me-15); ¹³C nmr (CDCl₃) δ 39.8 (C-1), 22.6 (C-2), 75.1 (C-3), 31.7 (C-4), 49.5 (C-5), 79.7 (C-6), 53.6 (C-7), 23.5 (C-8), 43.2 (C-9), 35.9 (C-10), 41.8 (C-11), 179.4 (C-12), 12.6 (C-13), 21.1 (C-14), 9.4 (C-15), 170.3 (MeCO), 21.3 (MeCO); cims (CH₄) m/z 295 [M+1]⁺ (10), 235 [M+1-60]⁺ (100); anal., found C 69.3, H 8.9; C₁₇H₂₆O₄ requires C 69.34, H 8.91%.

 3β -Acetoxy-4 α -bydroxy-5 α , 11 β -H-eudesman-6 α , 12-olide [13].—Mp 197°; [α]²⁵D + 14.5°(c=1, CHCl₃); ir ν max (CHCl₃) 3565, 1781, 1736 cm⁻¹; ¹H nmr (CHCl₃) δ 4.74 (1H, ddd, $J_{3,2\beta}$ =11.5 Hz, $J_{3,2\alpha}$ = $J_{3,4\alpha}$ =4.9 Hz, H-3 α), 4.05 (1H, dd, $J_{6,5}$ =11.3 Hz, $J_{6,7}$ =10.3 Hz, H-6 β), 2.26 (1H, dq, $J_{11,7}$ =12.5 Hz, $J_{11,13}$ =6.9 Hz, H-11 β), 2.06 (3H, s, Me-AcO), 1.20 (3H, d, $J_{13,11}$ =6.9 Hz, Me-13), 1.33 (3H, s, Me-15), 0.99 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 39.1 (C-1), 23.5 (C-2), 78.8 (C-3), 73.5 (C-4), 56.3 (C-5), 80.4 (C-6), 53.3 (C-7), 24.8 (C-8), 42.8 (C-9), 37.4 (C-10), 40.5 (C-11), 178.2 (C-12), 12.5 (C-13), 21.4 (C-14), 19.0 (C-15), 170.7 (MeCO), 19.9 (MeCO); cims (CH₄) m/z 311 [M+1]⁺ (42), 293 [M+1-18]⁺ (76), 251 [M+1-60]⁺ (100); anal., found C 65.8, H 8.5; C₁₇H₂₆O₅ requires C 65.77, H 8.45%.

3 β ,8 α -Diacetoxy-4 α ,5 α ,11 β -H-eudesman-6 α ,12-olide [14].—Mp 203°, [α]²⁵D +12° (c=1, CHCl₃); ir ν max (CHCl₃) 1780, 1733, 1712 cm⁻¹; ¹H nmr (CDCl₃) δ 5.08 (1H, ddd, $J_{8\beta,7}$ =15.1 Hz, $J_{8\beta,9\alpha}$ =10.9 Hz, $J_{8\beta,9\beta}$ =4.2 Hz, H-8 β), 4.81 (1H, ddd, $J_{3,2\beta}$ =10.7 Hz, $J_{3,2\alpha}$ = $J_{3,4\alpha}$ =5.2 Hz, H-3 α), 4.00 (1H, dd, $J_{6,5}$ =11.2 Hz, $J_{6,7}$ =10.9 Hz, H-6 β), 2.55 (1H, dq, $J_{11,7}$ =12.5 Hz, $J_{11,13}$ =6.9 Hz, H-11 β), 2.04 (3H, s, Me-AcO), 2.03 (3H, s, Me-AcO), 1.22 (3H, d, $J_{13,11}$ =6.9 Hz, Me-13), 1.11 (3H, s, Me-14), 0.96 (3H, d, $J_{15,4}$ =7.5 Hz, Me-15); ¹³C nmr (CDCl₃) δ 39.4 (C-1), 22.3 (C-2), 74.7 (C-3), 31.6 (C-4), 49.0 (C-5), 76.8 (C-6), 57.1 (C-7), 70.2 (C-8), 49.1 (C-9), 35.1 (C-10), 41.2 (C-11), 178.3 (C-12), 14.0 (C-13), 21.8 (C-14), 9.4 (C-15), 170.4 (MeCO), 170.4 (MeCO), 21.3 (MeCO), 21.1 (MeCO); cims (CH₄) m/z 353 [M+1]⁺ (29), 293 [M+1-60]⁺ (64), 233 [M+1-120]⁺ (100); anal., found C 64.7, H 7.9; C₁₉H₂₈O₆ requires C 64.74, H 8.01%.

3β-Acetoxy-11β-bydroxy-4α, 5α-H-eudesman-6α, 12-olide [15].—Syrup; $\{\alpha\}^{25}$ D +13° (c=1, CHCl₃); ir ν max (CHCl₃) 3449, 1783, 1734 cm⁻¹; ¹H nmr (CDCl₃) δ 4.80 (1H, ddd, $J_{3,28}$ =10.3 Hz, $J_{3,2α}$ = $J_{3,4α}$ =4.9 Hz, H-3α), 4.43 (1H, dd, $J_{6,5}$ =11.7 Hz, $J_{6,7}$ =10.0 Hz, H-6β), 2.03 (3H, s, Me-AcO), 1.24 (3H, s, Me-13), 1.05 (3H, s, Me-14), 1.00 (3H, d, $J_{15,4}$ =7.4 Hz, Me-15); ¹³C nmr (CDCl₃) δ 39.9 (C-1), 22.6 (C-2), 75.1 (C-3), 31.8 (C-4), 49.8 (C-5), 78.3 (C-6), 55.9 (C-7), 18.3 (C-8), 43.0 (C-9), 35.9 (C-10), 74.2 (C-11), 177.8 (C-12), 21.8 (C-13), 21.0 (C-14), 9.5 (C-15), 170.5 (MeCO), 21.3 (MeCO); cims (CH₄) m/z 311 [M+1]⁺ (7), 233 [M+1-78]⁺ (56); anal., found C 65.6, H 8.3; C₁₇H₂₆O₅ requires C 65.77, H 8.45%.

BIOTRANSFORMATION OF 3β -HYDROXY- 4β , 5α , 11β -H-EUDESMAN- 6α ,12-OLIDE [4].—Substrate 4 (800 mg) was dissolved in EtOH (10 ml), distributed among 18 Erlenmeyer-flask cultures and incubated for 10 days, after which the cultures were processed as indicated above for the biotransformation of substrate 2 to give a mixture (785 mg) that was chromatographed on a Si gel column to obtain 72 mg(8.5%) of starting material, 344 mg (40%) of metabolite 16, 64 mg (7.5%) of metabolite 17, 136 mg (16%) of metabolite 18, and 120 mg (13%) of metabolite 19.

 $1\beta_{,3}\beta_{-Dibydroxy-4}\beta_{,5\alpha,11-H-eudesman-6\alpha,12-olide} [16]_{---Syrup;} [\alpha]^{25}D + 39^{\circ} (c=1, CHCl_{3}); ir \nu max (CHCl_{3}) 3408, 1759 cm^{-1}; {}^{1}H nmr (CDCl_{3}) \delta_{,3.92} (1H, dd, J_{6,5}=J_{6,7}=10.5 Hz, H-6\beta), 3.37 (1H, dd, J_{1\alpha,2\beta}=11.9 Hz, J_{1\alpha,2\alpha}=4.4 Hz, H-1\alpha), 3.19 (1H, ddd, J_{3,2\beta}=11.4 Hz, J_{3,4\alpha}=9.8 Hz, J_{3,2\alpha}=5.1 Hz, H-3\alpha), 2.21 (1H, dq, J_{11,7}=12.3 Hz, J_{11,13}=6.9 Hz, H-11\beta), 1.19 (3H, d, J_{13,11}=6.9 Hz, Me-13), 1.19 (3H, d, J_{15,4}=6.4 Hz, Me-15), 0.95 (3H, s, Me-14); {}^{13}C nmr (CDCl_{3}) \delta_{,7.4} (C-1), 36.8 (C-2), 76.1 (C-3), 38.5 (C-4), 50.3 (C-5), 82.7 (C-6), 53.7 (C-7), 23.1 (C-8), 39.2 (C-9), 42.1 (C-10), 40.7 (C-11), 179.4 (C-12), 12.3 (C-13), 12.5 (C-14), 16.4 (C-15); cims (CH₄) m/z 269 [M+1]⁺ (28), 251 [M+1-18]⁺ (26), 233 [M+1-36]⁺ (47); anal., found C 67.2, H 9.1; C_{1,2}A_{4}A_{4} requires C 67.12, H 9.02\%.$

3β,8α-Dibydroxy-4β,5α,11β-H-eudesman-6α,12-olide [17].—Mp 201°, [α]²⁵D -18° (c=1, CHCl₃); ir ν max (CHCl₃) 3417, 1761 cm⁻¹; ¹H nmr (CDCl₃) δ 3.94 (1H, ddd, $J_{38,7}$ =14.9 Hz, $J_{88,96}$ =10.6 Hz, $J_{88,96}$ =4.4 Hz, H-8β), 3.83 (1H, dd, $J_{68,7}$ = $J_{68,5}$ =10.8 Hz, H-6β), 3.15 (1H, ddd, $J_{3,28}$ =10.9 Hz, $J_{3,4\alpha}$ =9.6 Hz, $J_{3,2\alpha}$ =5.1 Hz, H-3α), 2.48 (1H, dq, $J_{11,7}$ =13.8 Hz, $J_{11,13}$ =6.9 Hz, H-11β), 1.36 (3H, d, $J_{13,11}$ =6.9 Hz, Me-13), 1.20 (3H, d, $J_{15,4}$ =6.3 Hz, Me-15), 0.98 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 39.7 (C-1), 29.9 (C-2), 76.0 (C-3), 38.6 (C-4), 51.8 (C-5), 80.6 (C-6), 59.8 (C-7), 68.9 (C-8), 51.3 (C-9), 36.3 (C-10), 40.6 (C-11), 179.1 (C-12), 14.5 (C-13), 16.5 (C-14), 20.0 (C-15); cims (CH₄) m/z 269 [M+1]⁺ (46), 251 [M+1-18]⁺ (69), 233 [M+1-36]⁺ (100); anal., found C 67.1, H 8.9; C₁₅H₂₄O₄ requires C 67.12, H 9.02%.

 $3\beta_{,9}\beta$ -Dihydroxy-4 $\beta_{,5}\alpha_{,11}\beta$ -H-eudesman-6 $\alpha_{,12}$ -olide [18].—Mp 194°; [α]²⁵D +74° (c=1, CHCl₃); ir ν max (CHCl₃) 3419, 1768 cm⁻¹; ¹H nmr (CDCl₃) δ 3.81 (1H, dd, $J_{6,5}=J_{6,7}=10.5$ Hz, H-6 β), 3.37 (1H, dd, $J_{9,8\beta}=10.2$ Hz, $J_{9,8\alpha}=4.2$ Hz, H-9 α), 3.12 (1H, ddd, $J_{3,2\beta}=11.3$ Hz, $J_{3,4\alpha}=9.8$ Hz, $J_{3,2\alpha}=4.7$ Hz, H-3 α), 2.26 (1H, dq, $J_{11,7}=12.6$ Hz, $J_{11,13}=6.8$ Hz, H-11 β), 1.20 (3H, d, $J_{15,4}=6.4$ Hz, Me-15), 1.19 (3H, d, $J_{13,11}=6.8$ Hz, Me-13), 0.97 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 36.0 (C-1), 31.6 (C-2), 75.8 (C-3), 38.4 (C-4), 50.3 (C-5), 82.4 (C-6), 51.0 (C-7), 30.1 (C-8), 78.1 (C-9), 41.5 (C-10), 40.5 (C-11), 179.1 (C-12), 12.5 (C-13), 12.8 (C-14), 16.5 (C-15); cims (CH₄) m/z 269 [M+1]⁺ (35), 251 [M+1-18]⁺ (47), 233 [M+1-36]⁺ (100); anal., found C 67.1, H 9.0; C₁₅H₂₄O₄ requires C 67.12, H 9.02%.

 $3\beta, 11\beta$ -Dibydroxy-4 $\beta, 5\alpha, 11\beta$ -H-eudesman- $6\alpha, 12$ -olide [**19**].—Mp 187°; [α]²⁵D + 62° (c=1, CHCl₃); ¹H nmr (CDCl₃) δ 4.29 (1H, dd, $J_{6,5}$ =11.0 Hz, $J_{6,7}$ =10.1 Hz, H-6 β), 3.13 (1H, ddd, $J_{3,2\beta}$ =10.9 Hz, $J_{3,4\alpha}$ =9.9 Hz, $J_{3,2\alpha}$ =5.0 Hz, H-3 α), 1.42 (3H, s, Me-13), 1.20 (3H, d, $J_{15,4}$ =6.4 Hz, Me-15), 0.99 (3H, s, $\begin{array}{l} \textbf{Me-14}; \ ^{13}C \ nmr \ (\textbf{CDCl}_3) \ \& \ 40.0 \ (\textbf{C-1}), \ 30.4 \ (\textbf{C-2}), \ 76.2 \ (\textbf{C-3}), \ 39.0 \ (\textbf{C-4}), \ 52.8 \ (\textbf{C-5}), \ 82.0 \ (\textbf{C-6}), \ 56.0 \ (\textbf{C-7}), \ 17.9 \ (\textbf{C-8}), \ 40.7 \ (\textbf{C-9}), \ 36.9 \ (\textbf{C-10}), \ 77.3 \ (\textbf{C-11}), \ 179.5 \ (\textbf{C-12}), \ 21.6 \ (\textbf{C-13}), \ 18.8 \ (\textbf{C-14}), \ 16.7 \ (\textbf{C-15}); \ cims \ (\textbf{CH}_4) \ m/z \ 269 \ [\textbf{M+1]}^+ \ (38), \ 251 \ [\textbf{M+1-18]}^+ \ (65), \ 233 \ [\textbf{M+1-36]}^+ \ (100); \ \textit{anal.}, \ found \ \textbf{C} \ 67.2, \ \textbf{H} \ 9.0; \ \textbf{C}_{15}\textbf{H}_{24}\textbf{O}_4 \ requires \ \textbf{C} \ 67.12, \ \textbf{H} \ 9.02\%. \end{array}$

BIOTRANSFORMATION OF 3α -HYDROXY- 4β , 5α , 11β -H-EUDESMAN- 6α ,12-OLIDE [5].—Substrate 5 (220 mg) was dissolved in EtOH (4 ml), distributed among four Erlenmeyer-flask cultures and incubated for 10 days, after which the cultures were processed as indicated above for the biotransformation of substrate 2. Chromatography on Si gel yielded 33 mg (15%) of starting material, 55 mg (30%) of metabolite 16, which was also isolated in the biotransformation of substrate 4, and 40 mg (17%) of metabolite 20.

 $3\alpha,9\beta$ -Dihydroxy-4 $\beta,5\alpha,11\beta$ -H-eudesman-6 $\alpha,12$ -olide [20].—Syrup; [α]²⁵D +9° (z=1, CHCl₃); ir ν max (CHCl₃) 3443, 1762 cm⁻¹; ¹H nmr (CDCl₃) δ 3.78 (1H, dd, $J_{6,5}$ =11.0 Hz, $J_{6,7}$ =10.0 Hz, H-6 β), 3.76 (1H, m, H-3 β), 3.47 (1H, dd, $J_{9,8\beta}$ =10.6 Hz, $J_{9,8\alpha}$ =4.5 Hz, H-9 α), 2.25 (1H, dq, $J_{11,7}$ =12.2 Hz, $J_{11,13}$ =6.9 Hz, H-11 β), 1.19 (3H, d, $J_{13,11}$ =6.9 Hz, Me-13), 1.15 (3H, d, $J_{15,4}$ =6.8 Hz, Me-15), 0.95 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 31.5 (C-1), 28.5 (C-2), 71.9 (C-3), 34.8 (C-4), 45.1 (C-5), 82.3 (C-6), 50.2 (C-7), 31.3 (C-8), 78.2 (C-9), 41.7 (C-10), 40.7 (C-11), 179.4 (C-12), 12.5 (C-13), 11.8 (C-14), 18.0 (C-15); cims (CH₄) m/z 269 [M+1]⁺ (20), 251 [M+1-18]⁻ (25), 233 [M+1-36]⁺ (61); anal., found C 67.2, H 9.0; C₁₃H₂₄O₄ requires C 67.12, H 9.02%.

BIOTRANSFORMATION OF 3 β -HYDROXY-4 α ,5 α ,11 β -H-EUDESMAN-6 β ,12-OLIDE [7].—Substrate 7 (1 g) was dissolved in EtOH (20 ml), distributed among 20 Erlenmeyer-flask cultures and incubated for 12 days, after which the cultures were processed as indicated above for the biotransformation of substrate **2**. The resulting mixture (800 mg) was chromatographed on a Si gel column to obtain 100 mg (10%) of starting material, 165 mg (16%) of metabolite **21**, 110 mg (20%) of metabolite **22** (18), 130 mg (12%) of metabolite **23**, and 60 mg of a polar mixture which was dissolved in Ac₂O/pyridine (1:2) and maintained for 10 h at room temperature. The reaction mixture was diluted with H₂O, extracted with CH₂Cl₂, washed with saturated aqueous KHSO₄ and dried with anhydrous Na₂SO₄. Cc on Si gel yielded acetoxy derivatives **24** (4%), **25** (10%), and **26** (18) (5%). All yields are given with respect to substrate **7**.

 $\begin{array}{l} 1\beta, 3\beta - Dibydroxy - 4\alpha, 5\alpha, 11\beta - H-eudesman-6\beta, 12-olide [21]. \\ --Syrup; [\alpha]^{25}D - 77.5^{\circ}(c=1, CHCl_3); ir \\ \nu \max (CHCl_3) 3373, 1768 cm^{-1}; {}^{1}H nmr (CDCl_3) \delta 4.60 (1H, dd, J_{6,7}=4.1 Hz, J_{6,5}=3.0 Hz, H-6\alpha), 3.78 \\ (1H, ddd, J_{3,2\beta}=11.7 Hz, J_{3,2\alpha}=5.2 Hz, J_{3,4\alpha}=4.5 Hz, H-3\alpha), 3.21 (1H, dd, J_{1\alpha,2\beta}=11.5 Hz, J_{1\alpha,2\alpha}=4.3 Hz, \\ H-1\alpha), 2.32 (1H, q, J_{11,13}=7.6 Hz, H-11\beta), 1.27 (3H, d, J_{13,11}=7.6 Hz, Me-13), 1.19 (3H, d, J_{15,4}=7.4 Hz, \\ Me-15), 1.06 (3H, s, Me-14); {}^{13}C nmr (CDCl_3) \delta 78.0 (C-1), 35.5 (C-2), 71.3 (C-3), 40.2 (C-4), 45.2 (C-5), 82.2 (C-6), 43.7 (C-7), 23.8 (C-8), 38.5 (C-9), 37.7 (C-10), 43.5 (C-11), 180.2 (C-12), 14.1 (C-13), 15.4 \\ (C-14), 9.7 (C-15); cims (CH_4) m/z 269 [M+1]^+ (21), 251 [M+1-18]^+ (100), 233 [M+1-36]^+ (69); \\ anal., found C 67.1, H 8.9; C_{15}H_{24}O_4 requires C 67.12, H 9.02\%. \end{array}$

3 β ,8 α -Dibydroxy-4 α ,5 α ,11 β -H-eudesman-6 β ,12-olide [23].—Mp 149°; [α]²⁵D -60.0° (c=1,CHCl₃); ir ν max (CHCl₃) 3585, 1753 cm⁻¹; ¹H nmr (CDCl₃) δ 4.70 (1H, dd, $J_{6,5}$ =2.9 Hz, $J_{6,7}$ =4.5 Hz, H-6 α), 3.68 (1H, ddd, $J_{3,2\beta}$ =9.6 Hz, $J_{3,2\alpha}$ = $J_{3,4\alpha}$ =4.7 Hz, H-3 α), 3.53 (1H, ddd, $J_{8\beta,7}$ =12.0 Hz, $J_{8\beta,9\alpha}$ =9.9 Hz, $J_{8\beta,9\alpha}$ =4.2 Hz, H-8 β), 2.69 (1H, q, $J_{11,13}$ =7.6 Hz, H-11 β), 1.86 (1H, dd, $J_{7,8\beta}$ '=9.9 Hz, $J_{7,6}$ =4.5 Hz, H-7), 1.25 (3H, d, $J_{13,11}$ =7.6 Hz, Me-13), 1.11 (3H, d, $J_{15,4}$ =7.4 Hz, Me-15), 1.06 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 41.4 (C-1), 26.2 (C-2), 74.1 (C-3), 41.6 (C-4), 47.7 (C-5), 84.3 (C-6), 52.7 (C-7), 67.6 (C-8), 52.5 (C-9), 34.5 (C-10), 40.6 (C-11), 180.1 (C-12), 14.4 (C-13), 22.5 (C-14), 10.3 (C-15); cims (CH₄) m/z 269 [M+1]⁺ (33), 251 [M+1-18]⁺ (100), 233 [M+1-36]⁺ (87); anal., found C 67.0, H 8.9; C₁₅H₂₄O₄ requires C 67.12, H 9.02%.

 $1\beta-Acetoxy-3\beta-bydroxy-4\alpha, 5\alpha, 11\beta-H-eudesman-6\beta, 12-olide [24]. \\ --Syrup; [\alpha]^{25}D - 88.6^{\circ} (c=1, CHCl_3); ir \nu max (CHCl_3) 3461, 1768, 1728 cm⁻¹; ¹H nmr (CDCl_3) \delta 4.67 (1H, dd, <math>J_{1\alpha,2\beta} = 11.3 \text{ Hz}, J_{1\alpha,2\alpha} = 4.3 \text{ Hz}, H-1\alpha), 4.61 (1H, dd, <math>J_{6,7} = 4.3 \text{ Hz}, J_{6,5} = 2.8 \text{ Hz}, H-6\alpha), 3.84 (1H, ddd, <math>J_{3,2\beta} = 11.3 \text{ Hz}, J_{3,2\alpha} = J_{3,4\alpha} = 5.1 \text{ Hz}, H-3\alpha), 2.03 (3H, s, Me-AcO), 1.27 (1H, q, J_{11,13} = 7.6 \text{ Hz}, H-11\beta), 1.22 (3H, d, J_{15,4} = 7.4 \text{ Hz}, Me-15), 1.13 (3H, s, Me-14); ¹³C nmr (CDCl_3) \delta 78.7 (C-1), 29.8 (C-2), 70.9 (C-3), 40.0 (C-4), 45.1 (C-5), 82.0 (C-6), 43.5 (C-7), 23.6 (C-8), 38.3 (C-9), 36.9 (C-10), 43.5 (C-11), 180.3 (C-12), 14.0 (C-13), 16.4 (C-14), 9.7 (C-15), 170.1 (MeCO), 21.2 (MeCO); cims (CH_4) m/z 311 [M+1]⁺ (48), 293 [M+1-18]⁺ (24), 251 [M+1-60]⁺ (100), 233 [M+1-78]⁺ (87); anal., found C 65.8, H 8.5; C_{17}H_{26}O_5 requires C 65.77, H 8.45\%.$

 $_{3\beta-Acetoxy-4\alpha-bydroxy-5\alpha, 11\beta-H-eudesman-6\beta, 12-olide$ [25].—Mp 174°; [α]²⁵D – 39° (c=1, CHCl₃); ir ν max (CHCl₃) 3478, 1770 cm⁻¹; ¹H nmr (CDCl₃) δ 5.15 (1H, dd, $J_{6,7\alpha}$ =4.0 Hz, $J_{6,5}$ =2.6 Hz, H-6 α), 4.61 (1H, dd, $J_{3,2\beta}$ =11.9 Hz, $J_{3,2\alpha}$ = $J_{3,4\alpha}$ =4.7 Hz, H-3 α), 2.30 (1H, q, $J_{11,13}$ =7.6 Hz, H-11 β), 2.10 (3H, s, Me-AcO), 1.43 (3H, s, Me-15), 1.28 (3H, d, $J_{13,11}$ =7.6 Hz, Me-13), 1.06 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 39.4 (C-1), 23.8 (C-2), 82.0 (C-3), 74.2 (C-4), 53.6 (C-5), 75.7 (C-6), 43.6 (C-7), 25.2 (C-8), 42.4 (C-9), 33.5 (C-10), 43.1 (C-11), 180.5 (C-12), 14.1 (C-13), 20.4 (C-14), 19.4 (C-15), 171.8 (MeCO), 21.4 (MeCO); cims (CH₄) m/z 311 [M+1]⁺ (15), 293 [M+1-18]⁺ (100), 251 [M+1-60]⁺ (81), 233 [M+1-78]⁺ (46); anal., found C 65.7, H 8.5; C₁₇H₂₆O₅ requires C 65.77, H 8.4%.

BIOTRANSFORMATION OF 3α -HYDROXY-4 β , 5α ,11 β -H-EUDESMAN-6 β ,12-OLIDE [9].—Substrate 9 (450 mg) was dissolved in EtOH (10 ml), distributed among 10 Erlenmeyer-flask cultures and incubated for 7 days, after which the cultures were throughly processed as indicated above for the biotransformation of substrate 2. This yielded a mixture (420 mg) that was chromatographed on Si gel to afford 40 mg (9%) of starting material, 75 mg (17%) of product 8, 150 mg (31%) of metabolite 27, and 200 mg of a polar mixture that was dissolved in Ac₂O-pyridine (1:2) and maintained for 14 h at room temperature. Chromatography on Si gel yielded 40 mg (4%) of product 28, 40 mg (4%) of product 29, and 100 mg (17%) of product 30. All yields are given with respect to substrate 9.

3 α ,8 α -Dibydroxy-4 β ,5 α ,11 β -H-eudesman-6 β ,12-olide [27].—Mp 174°; { α }]²⁵D -36° (c=1, CHCl₃); ¹H nmr (CDCl₃) δ 4.72 (1H, dd, $J_{6,7}$ =5.2 Hz, $J_{6,5\alpha}$ =3.7 Hz, H-6 α), 3.90 (1H, dd, J_3 =2.6 Hz, J_1 = J_2 =2.7 Hz, H-3 β), 3.70 (1H, ddd, $J_{8,7}$ =12 Hz, $J_{8,9\alpha}$ =10.1 Hz, $J_{8,9\alpha}$ =4.5 Hz, H-8 β), 2.82 (1H, q, $J_{11,13}$ =7.7 Hz, H-11 β), 1.34 (3H, d, $J_{13,11}$ =7.7 Hz, Me-13), 1.07 (3H, d, $J_{15,4}$ =6.8 Hz, Me-15), 0.97 (3H, s, Me-14); ¹³C nmr (CDCl₃) δ 36.0 (C-1), 28.8 (C-2), 71.9 (C-3), 33.1 (C-4), 42.9 (C-5), 79.1 (C-6), 51.3 (C-7), 68.1 (C-8), 48.9 (C-9), 34.6 (C-10), 42.1 (C-11), 180.3 (C-12), 15.0 (C-13), 18.6 (C-14), 15.5 (C-15); cims (CH₄) m/z 269 [M+1]⁺ (16), 251 [M+1-18]⁺ (89), 233 [M+1-36]⁺ (100); anal., found C 67.0, H 9.0; C₁₃H₂₄O₄ requires C 67.12, H 9.02%.

 1β -Acetoxy- 3α -bydroxy- 4β , 5α , 11β -H-eudesman- 6α , 12-olide [28].—Mp 152°; $[\alpha]^{25}D - 50°$ (c=1, CHCl₃); ir ν max (CHCl₃) 3496, 1767, 1737 cm⁻¹; ¹H nmr (CDCl₃) δ 4.86 (1H, dd, $J_{1,2\alpha}$ =11.7 Hz, $J_{1,2\beta}$ =5.0 Hz, H-1 α), 4.60 (1H, dd, $J_{6,5\alpha}$ = $J_{6,7}$ =6.9 Hz, H-6 α), 3.98 (1H, dd, J_3 =2.7 Hz, J_1 = J_2 =2.8 Hz, H-3 β), 2.35 (1H, q, $J_{11,13}$ =7.7 Hz, H-11 β), 2.01 (3H, s, Me-AcO), 1.28 (3H, d, $J_{13,11}$ =7.7 Hz, Me-13), 1.07 (3H, d, $J_{15,4}$ =6.8 Hz, Me-15), 1.00 (3H, s, Me-14); ¹³C nmr (CDCl₃) 77.4 (C-1), 34.8 (C-2), 71.3 (C-3), 32.9 (C-4), 41.9 (C-5), 76.3 (C-6), 41.8 (C-7), 23.1 (C-8), 34.8 (C-9), 37.1 (C-10), 44.5 (C-11), 180.4 (C-12), 14.6 (C-13), 12.6 (C-14), 14.9 (C-15), 170.8 (MeCO), 21.3 (MeCO); cims (CH₄) m/z 311 [M+1]⁺ (16), 251 [M+1-60]⁺ (56); anal., found C 65.6, H 8.4; C₁₇H₂₆O, requires C 65.77, H 8.45%.

1β, 3α-Diacetoxy-4β, 5α, 11β-H-eudesman-6α, 12-olide [29].—Mp 175°; $[α]^{25}D - 26°(c=1, CHCl_3)$; ir ν max (CHCl_3) 1771, 1736 cm⁻¹; ¹H nmr (CDCl_3) δ 5.14 (1H, dd, $J_1=J_2=3.0$ Hz, $J_3=2.9$ Hz, H-3β), 4.77 (1H, dd, $J_{1,2a}=11.7$ Hz, $J_{1,2B}=4.9$ Hz, H-1α), 4.60 (1H, dd, $J_{6,5a}=J_{6,7}=4.0$ Hz, H-6α), 2.38 (1H, q, $J_{11,13}=7.5$ Hz, H-11), 2.07 (3H, s, Me-AcO), 2.01 (3H, s, Me-AcO), 1.29 (3H, d, $J_{13,11}=7.5$ Hz, Me-13), 1.00 (3H, s, Me-14), 0.98 (3H, d, $J_{15,4}=6.7$ Hz, Me-15); ¹³C nmr (CDCl₃) δ 76.8 (C-1), 31.7 (C-2), 73.6 (C-3), 31.7 (C-4), 43.2 (C-5), 76.0 (C-6), 41.9 (C-7), 23.0 (C-8), 34.7 (C-9), 37.0 (C-10), 44.5 (C-11), 180.1 (C-12), 14.4 (C-13), 12.6 (C-14), 14.6 (C-15), 170.6 (MeCO), 170.5 (MeCO), 21.2 (MeCO); cims (CH₄) m/z 353 [M+1]⁺ (57), 293 [M+1-60]⁺ (76), 233 [M+1-120]⁺ (100); anal., found C 64.7, H 8.0; C₁₉H₂₈O₆ requires C 64.74, H 8.01%.

1β, 3β-Diacetoxy-4β, 5α, 11β-H-eudesman-6β, 12-olide [**30**].—Syrup; $[α]^{2^5}D - 15^\circ (c=1, CHCl_3)$; ir ν max (CHCl_3) 1771, 1739 cm⁻¹; ¹H nmr (CDCl_3) δ 4.70 (1H, dd, $J_{6,5\alpha} = J_{6,7} = 4.0$ Hz, H-6α), 4.53 (1H, dd, $J_{1,2\alpha} = 12.2$ Hz, $J_{1,2\beta} = 4.6$ Hz, H-1α), 4.51 (1H, dd, $J_1 = J_2 = 12.1$ Hz, $J_3 = 5.4$ Hz, H-3α), 2.38 (1H, qd, $J_{11,13} = 7.7$ Hz, H-11β), 2.16 (3H, s, Me-AcO), 2.01 (3H, s, Me-AcO), 1.29 (3H, d, $J_{13,11} = 7.7$ Hz, Me-13), 1.03 (3H, s, Me-14), 0.97 (3H, d, $J_{13,4\beta} = 6.4$ Hz, Me-15); ¹³C nmr (CDCl₃) δ 76.9 (C-1), 32.2 (C-2), 74.6 (C-3), 33.3 (C-4), 45.9 (C-5), 75.0 (C-6), 41.7 (C-7), 22.7 (C-8), 34.6 (C-9), 36.4 (C-10), 44.2 (C-11), 179.9 (C-12), 14.3 (C-13), 13.3 (C-14), 14.4 (C-15), 170.4 (MeCO), 170.2 (MeCO), 21.0 (MeCO), 20.9 (MeCO); cims (CH₄) m/z 353 [M+1]⁺ (29), 293 [M+1-60]⁺ (100), 233 [M+1-120]⁺ (65); anal., found C 64.6, H 8.0; C₁₉H₂₈O₆ requires C 64.74, H 8.01%.

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